Fabrication of Ti₂AIC-Ti₃AIC₂-Ti₃SiC₂ composite by spark plasma sintering (SPS) method from elemental powders

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Recently, a new remarkable group of materials known as ternary-layered compounds or "machinable ceramics" has attracted interest of materials scientists. These ternary-layered ceramics are thermally and electrically conductive, relatively soft and readily machinable, they are relatively tough and not susceptible to thermal shock, and behave plastically at elevated temperatures. At the same time, these carbides are very refractory, oxidation resistant and maintain strength to temperatures that render the best superalloys available today unusable [1]. This unique combination of properties makes them likely candidates for structural applications at elevated temperatures, such as turbine blades and stators, heavy duty electric contacts, bearings, etc.

 $Ti₂AIC$ and $Ti₃AIC₂$ in Ti-Al-C system and $Ti₃SiC₂$ in Ti-Si-C system are typical compounds belonging to this family. However, it is difficult to fabricate single-phase bulk dense samples of $Ti₂AIC$, $Ti₃AIC₂$ or $Ti₃SiC₂$ because of their very narrow phase range in Ti-Al-C and Ti-Si-C phase diagrams. The resultant samples always contains, in most cases, TiC, as ancillary unwanted phase [2–4]. It was also found that small amount of Ti₂AlC always exists in Ti₃AlC₂ samples [5]. Furthermore, it has been proved that Al additive apparently enhances the synthesis of $Ti₃SiC₂$ [6]. When 0.3 mol Al was used to replace Si for preparing $Ti₃SiC₂$, a new unknown phase in addition to $Ti₃SiC₂$ appeared in the resultant product [7]. This new unknown phase is probably $Ti₃AIC₂$ based on our analysis of the X-ray diffraction pattern. These imply that it is easier to synthesize composite containing these ternary carbides than to make single-phase carbides. Moreover, $Ti₂AIC$, $Ti₃AIC₂$ and $Ti₃SiC₂$ have similar properties and composite consisting of them may also have properties alike. Additionally, Spark plasma sintering (SPS) method is an innovative technique for rapid sintering materials with finer grains and better properties than those prepared by existing sintering methods [8].

The objective of this work was to fabricate $Ti₂AIC Ti₃AIC₂-Ti₃SiC₂$ composite in Ti-Al-Si-C quaternary system by SPS method from elemental powders.

All of the work was conducted using powder mixture of titanium (99.0% pure, 10.6 μ m), Al (99.8%) pure, 12.8 μ m), Si (99.5% pure, 9.5 μ m) and carbon black (99%, 13.2 μ m) (all from Institute of Non-ferrous Metals, Beijing, China). In brief, the mixture with a designed composition was firstly mixed for 24 hr, then placed in a graphite die, 20 mm in diameter, and finally sintered in a spark plasma sintering system (Mode SPS-1050). The powder mixture was heated at a rate of 80° C/min until the requisite temperature was reached; the soaking time was 8 min.

In previous work, we endeavored in the fabrication of single phase $Ti₂AIC$, so the composition of the powder mixture in this work was designed as a basic composition of 2.0Ti/1.0Al/1.0C plus 0.2 mol Si and the final

Figure 1 X-ray diffraction patterns of samples sintered at (a) 1000 °C, (b) 1100 ◦C, (c) 1200 ◦C, and (d) 1300 ◦C from 2.0Ti/1.0Al/0.2Si/1.0C powder mixture by SPS.

Figure 2 Temperature dependence of vacuum, (a) sample sintered at 1200 °C and (b) sample sintered at 1300 °C.

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Figure 3 Microstructure of fracture surfaces of samples sintered at (a) 1100 °C, and (b) 1200 °C from 2.0Ti/1.0Al/0.2Si/1.0C powder mixture by SPS.

powder mixture was 2.0Ti/1.0Al/0.2Si/1.0C. Since Al and Si especially Al evaporate at high temperature, more $Al + Si$ than required by the stoichiometric of $Ti₂AIC$ was used to make up their loss.

Fig. 1 shows the phases presented after spark plasma sintering the 2.0Ti/1.0Al/0.2Si/1.0C powder mixture at different temperatures for 8 min, as determined by XRD. It is clear that when sintered at $1000\degree C$ for 8 min, peaks of Ti₂AlC, Ti₃AlC₂ and Ti₃SiC₂ appeared. But there existed other unwanted phases: TiC and Ti-Al intermetallics such as TiAl and $Ti₃Al$. When the sintering temperature increased to 1100 or 1200 \degree C, only peaks of Ti₂AlC, Ti₃AlC₂ and Ti₃SiC₂ appeared in samples, all unnecessary phases disappeared. The apparent differences between these two samples are: $Ti₂AIC$ content was more than that of $Ti₃AIC₂$ in sample sintered at $1100\,^{\circ}\text{C}$ and Ti_3AlC_2 content was more than that of Ti₂AlC in sample sintered at 1200 \degree C. While in sample sintered at 1300 \degree C, peaks of Ti₂AlC disappeared; peaks of TiC appeared again and $Ti₃AIC₂$ and $Ti₃SiC₂$ were other two major phases.

In Ti-Al-Si-C system, Al and Si are more volatile than other two elements. So, the appearance of TiC and the disappearance of Ti₂AlC in sample prepared at 1300 °C are possibly due to the rapid loss of Al and Si at high temperature. This is also verified by the temperature dependence of the vacuum pressure in the exuviated chamber of the SPS system, as shown in Fig. 2. In the spark plasma sintering process, the gas in the chamber is discharged out at a constant rate. As a result, the vacuum pressure maintains unchanged during the ordinary sintering process. However, a peak in the pressure may appear if a large quantity of gas is given out from a reaction in the sample. Pressure peaks appeared in both samples sintered at 1200 and 1300 °C. But for sample sintered at 1300 ℃, pressure peak was more pronounced than that of sample sintered at $1200\degree C$, which indicates the evaporation of Al or Si is more serious in sample sintered at 1300 $°C$, leading to the formation of TiC and the disappearance of $Ti₂AIC$.

Shown in Fig. 3 are the scanning electron micrographs of the fracture surfaces of samples prepared at 1100 and 1200 °C. For sample prepared at 1100 °C, two kinds of grains appeared, a plate-like one with grain size of 16 μ m and an equiaxed one with grain size of 5μ m. These two kinds of grains were distributed inhomogeneously. Sample synthesized at 1200 ◦C also had two kinds of grains. But these two kinds of grains had smaller difference in grain size in a range of $5-12 \mu m$ than that appeared in sample prepared at $1100\degree C$ and distributed more homogeneously. There were no pores in these two samples, showing that samples were dense.

It is concluded from X-ray diffraction patterns and scanning electron micrographs that dense bulk composite only containing $Ti₂AIC$, $Ti₃AIC₂$ and $Ti₃SiC₂$ could be conveniently synthesized by SPS from mixed elemental powders with small amount of Si additive at 1100 and 1200 $°C$, a temperature lower than those used in HP and HIP methods for preparing single phase ternary carbides materials [3, 9, 10]. The whole process is completely finished in a few min.

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